

## NOTES

## BACTERIAL METABOLISM OF DAUNOMYCIN

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As anthracycline antibiotics are almost universally antitumor agents, and several, including adriamycin and daunomycin, are useful clinically against neoplastic diseases<sup>1,2</sup>, it is worthwhile to develop analogs. One means to this end is the use of microbial biotransformation, and this communication discusses a limited application of this method to daunomycin.

Daunomycin (1) was used as a substrate in a conversion in which a crude enzyme preparation of *Streptomyces steffisburgensis* was used as the biological agent. The fermentation was extracted, and the isolated material was chromatographed to give two products identified as 7-deoxydaunomycinone (2) and 7-deoxydaunomycinol aglycone (3).

The principal product 2, which was obtained in a yield of 66%, was identified by comparison with an authentic sample by tlc in three solvent systems. The minor product 3, which was obtained in a yield of 31%, was identified in the same manner and also by a high resolution mass spectrum. It seems probable that 2 was formed

by a reductive glycosidic cleavage as was suggested previously for such reactions<sup>3,4</sup> as no hydrolysis products were detected.

It was found that conversion of 1 to 2 required DPNH (Fig. 1). Such a cofactor requirement has also been shown in the reductive cleavage of daunomycin by *Aeromonas hydrophila*. The further reduction of 2 to 3 involving C-13 carbonyl reduction is catalyzed by a TPNH linked keto reductase (Fig. 2). These reactions are outlined in Scheme 1. Evidence is not conclusive that 2 is the first product followed by carbonyl reduction to 3. However, the C-13 dihydrodaunomycin arising from initial reduction at C-13 was not found as a reaction product, and it was established that 2 can be converted to 3 making the indicated sequence probable.

In these experiments a further example of microbial metabolism of anthracyclines is presented which follows the same pathway as was previously demonstrated<sup>5-7</sup> for mammalian metabolism.

## Material and Methods

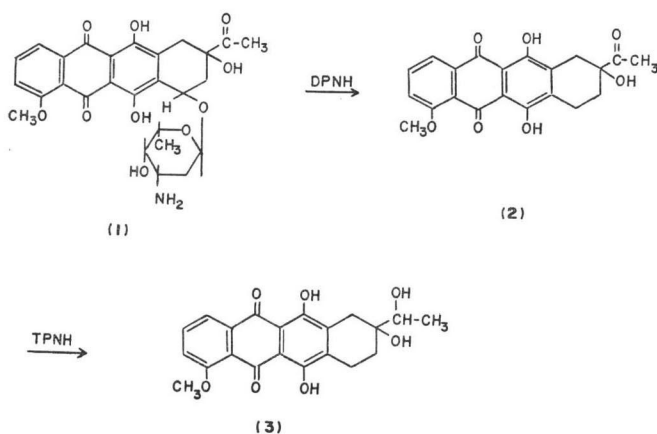
## Thin-layer systems

System 1: CHCl<sub>3</sub> - MeOH(95: 5)System 2: Cyclohexane - EtOAc - 95%EtOH  
(5: 3: 2)System 3: EtOAc - EtOH - H<sub>2</sub>O(92: 5: 3)

## Cell-free Extract Preparation

Five to 10 g of *S. steffisburgensis* grown aerobically for 48 hours at 28°C on tryptone - yeast extract - glucose (5: 3: 20) medium were suspended in 15 ml of 100 mM potassium phosphate buffer (pH 7.4) containing 11.7 mg (0.15 mmoles) of β-mercaptoethanol. The suspension was sonicated in the cold using a Raytheon 10 kc sonic oscillator followed by centrifugation at 10<sup>4</sup> × g for 15 minutes. The supernatant fluid was dialysed for 24 hours vs. 1 liter of the same buffer containing β-mercaptoethanol and was used as the crude enzyme source.

Scheme 1.



### Cell-free Conversion of Daunomycin (1)

An undialysed cell-free extract prepared from 25 g of *S. steffisburgensis* (containing an endogenous supply of TPNH) with a final volume of 40 ml was added to a solution of 20 ml of 1 M potassium phosphate (pH 7.4) and 10 ml of water. Fifty mg of daunomycin (1) dissolved in 1 ml of DMF was added. The solution was allowed to stand at 37°C for 20 hours.

The reaction mixture was mixed with 10 g of filter-aid and filtered, and the filter cake was washed with 50 ml of water. The filtrate was extracted with four 50-ml portions of  $\text{CH}_2\text{Cl}_2$ , and the filter cake was extracted with four 100-ml portions of  $\text{CH}_2\text{Cl}_2$ . The combined extracts were evaporated *in vacuo*, residue 125 mg, and chromatographed on silica (125 g) using  $\text{CHCl}_3$  - MeOH (99:1). The faster moving colored peak was combined as one pool and the slower moving as a second pool, and these were evaporated *in vacuo*. The less polar material gave 24 mg (66%) of material which had the same  $R_f$  as 7-deoxydaunomycinone (2) in tlc in three solvent systems. In solvent systems 1, 2 and 3 the  $R_f$  values were, respectively, 0.24, 0.64 and 0.68. The more polar fraction gave 11.5 mg (31%). On tlc this material showed a strong spot in three solvent systems having the same  $R_f$  values as did 7-deoxydaunomycinol aglycone (3). In solvent systems 1, 2 and 3  $R_f$  values were, respectively, 0.18, 0.55 and 0.59. There was also a weak spot present attributed to 2. The more polar material gave *m/e* 384.1199 (Required for  $\text{C}_{21}\text{H}_{20}\text{O}_7$ . 384.1209) *m/e* (TMS derivative) 672.

### Effects of Pyridine Nucleotides

The following reaction protocols were employed.

(a) Reductive Cleavage: Reaction mixtures were made on the basis of 1 ml volumes and were run under aerobic conditions at 37°C. Each reaction mixture contained 10 mg of dialysed crude enzyme preparation measured using the LOWRY protein determination, 250  $\mu\text{g}$  of daunomycin (1), 1 mg of DPNH, 100  $\mu\text{moles}$  of potassium phosphate solution (pH 7.4) and 10  $\mu\text{moles}$  of  $\beta$ -mercaptoethanol.

(b) Keto Reduction: A reaction protocol essentially identical to the one discussed above was employed with the following modifications. Dialysed crude enzyme was added at 1 mg/ml

with 7-deoxydaunomycinone (2) and TPNH added at 250  $\mu\text{g}/\text{ml}$  and 1 mg/ml, respectively. Both reactions were terminated by quick freezing in acetone-dry ice mixtures.

In the cases of both (a) and (b), control reactions were run containing all reaction components with the exception of reduced pyridine nucleotide.

The 1 ml reaction mixtures were subjected to

Fig. 1. Reductive conversion of daunomycin to 7-deoxydaunomycinone.

Reaction conditions are described in text.

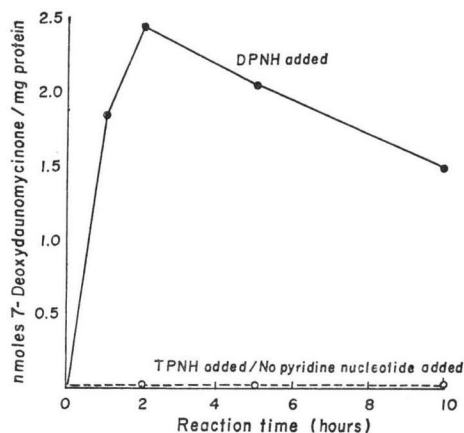
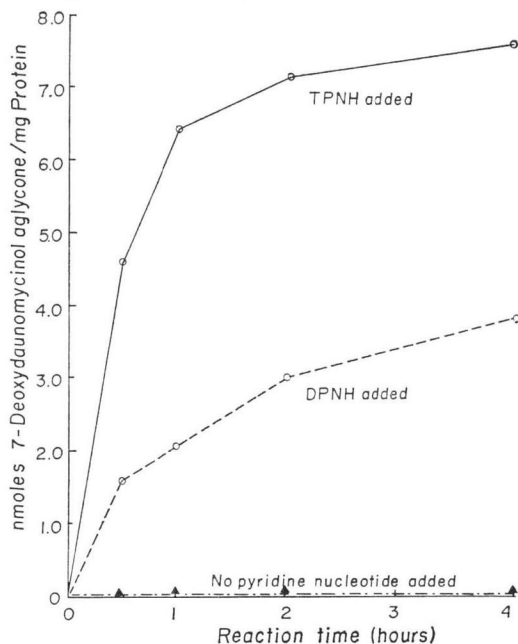


Fig. 2. Reduced pyridine nucleotide dependent conversion of 7-deoxydaunomycinone to 7-deoxydaunomycinol aglycone.

Reaction conditions are described in text.



mixing for 1 minute periods in the presence of 2 ml volumes of  $\text{CHCl}_3$ . The phases were separated by centrifugation for 5 minutes at  $5 \times 10^3 \times g$ . The contents of these extracts were quantitated by HPLC using a Chromatec 2200 unit equipped with a Hewlett Packard-Mosely 7128A recorder and a Hewlett Packard 3370B integrator. The absorbent was Partisil 10  $\mu$  PAC (Whatman), and the solvent was  $\text{CHCl}_3 - \text{CH}_3\text{OH} - \text{H}_2\text{O}$  (96:5:1). The anthracyclines, monitored at 254 nm, were quantitated on the basis of standard curves made using authentic compounds in a range of 1~4  $\mu\text{g}$ .

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